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Electronegativity greater
than 1.7 and is a transfer
of electrons. The Lower

Electronegativity transfers

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to the higher electronegativity) t. Ionic Bond. Ions pack in a crystal lattice structure. Lattice Energy.

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Test. Chapter: Chemical
Bonding. In the space
provided, write the letter
of the term or phrase that
best completes each.
statement or best answers
each question. _____ 1. The
charge on an ion is.
a. always positive. b. always
negative. c. either positive
or negative.

Assessment Chapter Test A - Kettering City School District

REVIEW Chemical Bonding
MIXED REVIEW SHORT ANSWER
Answer the following
questions in the space
provided. 1. a.

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Test Name Class Date Chapter
Test A, continued Use this
figure to answer questions 7
and 8. _____ 7. A solution
containing 35 g of Li_2SO_4
dissolved in 100 g of water
is heated from 10°C to 90°C .
According to information in

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the figure, this temperature change would result in a. an additional 5 g of Li_2SO_4 in ...

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Modern Chemistry 37 . Name
Section Quiz, continued
Class Date 6. Metals are
malleable because when
struck, one plane of metal
atoms can slide past another
plane without breaking
bonds. b. cannot easily move
out of the way. c. moves in
a way that maximizes the
repulsive forces within the

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Choice Identify the letter
of the choice that best
completes the statement or
answers the question. 1.
binds the atoms together is
called a(n) A mutual
electrical attraction
between the nuclei and
valence electrons of
different atoms that a.
dipole. c. chemical bond. b.
Lewis structure. d. London

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