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In this activity, students learn about solubility and concentration. In section 1, students answer questions from a video that explains the dissolving process and how to calculate the molarity of a solution.

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NEL 2.3 Concentration and Solubility 43 Figure 3 Which solution has had a lot of drink powder added to it? Suppose a solution contains 6.0 g of sugar in 200 mL of sugar-and-water solution. What is the concentration of the sugar-and-water solution? Given: mass of solute = 6.0 g volume of solution = 200 mL Required:

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concentration of the solution

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Reactions 197 Note The answer is about To calculate molar concentration we would need to "Test Method Protocol for Solubility Determination 2003 April 25th, 2018 - Test Method Protocol for Solubility Determination Phase III Note Suggested brand Section VII C 2 a If complete

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