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~~12=12 - solution~~

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How To Make 12% Annually ... Forever

When pH=12, [OH-] =0.01mol/L ? in one litre ?n=0.01mol ?M(NaOH)=40g/mol So 0.4g (m=n*M=0.4g)NaOH should be dissolved in one litre of solution. Hope these could help u~

To make a solution of pH 12, what should the amount of ...

18) What volume (in mL) of 0.0887 M Mg 12 solution is needed to make 275.0 mL of 0.0224 M Mg I2 solution? A) 72.3 mL. B) 144 mL C) 69.4 mL D)91.8 mL E) 10.9 mL

Solved: 18) What Volume (in ML) Of 0.0887 M Mg 12 Solution ...

For example, to make 100 ml of 0.1 M CaCl 2 solution, use the previous formula to find out how much CaCl 2 you need: grams of CaCl 2 = (0.1) x (110.91) x (100) ÷ (1000) = 1.11 g; Now you can make your solution: dissolve 1.11 g of CaCl 2 in sufficient water to make 100 ml of solution. The amount of water needed will be slightly less than 100 ml.

How to Make a Solution: Chemical, Molar and Weight Percent

Riley wants to make 100 mL of a 25% saline solution but only has access to 12% and 38% saline mixtures. Which of the following system of equations correctly describes this situation if X represents the amount of the 12% solution used, and y represents the amount of the 38% solution used? a.) 0.12x + 0.38y = 0.25(100) x + y = 100 b.) 0.38x + 0.12y = 100 x + y = 0.25(100) c.) 0.38x + 0.12y = 0.25(100) x + y = 100 O ...

Solved: Riley Wants To Make 100 ML Of A 25% Saline Solutio ...

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